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Chemistry 101 ANSWER KEY 1 REVIEW
QUESTIONS Chapter 8 Use only a
periodic table to answer the following
questions. 1. Write complete electron

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configuration for each of the following elements: a) Aluminum (Al) $1s^2 2s^2 2p^6 3s^2 3p^1$ b) Sulfur (S) $1s^2 2s^2 2p^6 3s^2 3p^4$ c) Manganese (Mn) $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^5$

Chemistry Chapter 8 Section Review Answer Key

CHAPTER 8 REVIEW Chemical Equations

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and Reactions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. b A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

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ANSWER Answer the following questions
in the space provided. 1. b A balanced
chemical equation represents all the
following except (a) experimentally
established facts. (b) the mechanism by

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which reactants combine to form products.

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CHAPTER 8 REVIEW Chemical Equations

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and Reactions SECTION 3 SHORT
ANSWER Answer the following questions
in the space provided. 1. List four metals
that will not replace hydrogen in an acid.
Choose from Cu, Ag, Au, Pt, Sb, Bi, and
Hg.

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CHAPTER 8 REVIEW Chemical Equations and Reactions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. b A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

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Modern Chemistry Chapter 8 Mixed Review Answers

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Zumdahl 8113 The first step always is to
draw a valid Lewis structure when
predicting molecular structure When
resonance is possible, only one of the
possible resonance structures is

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necessary to predict the correct structure because all resonance structures give the same structure The Lewis

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equation. precipitate. a concise representation of a chemical reaction using chemical.... a chemical equation that does not indicate the relative amount.... an equation in which each side has the same number of atoms of....

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The chemical composition of substance c.... What are the four observations that

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Physical change. Chemical change. Gas
is released ... Precipitate - insoluble solid
is produced...

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Chapter 8 Chemistry Matter and Change
Vocab. diatomic. single covalent bond.

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double covalent bond. seven diatomic molecules (BrINClHOF) two atoms of the same element. 1 bond- 2e. 2 pairs of e- are shared. hydrogen, oxygen, nitrogen, flourine, chlorine, bromine, iodine. diatomic. two atoms of the same element. single covalent bond.

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exists as individual mo....

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CHAPTER 8 REVIEW Chemical Equations
and Reactions SECTION 3 SHORT

ANSWER Answer the following questions
in the space provided. 1. List four metals
that will not replace hydrogen in an acid.
Choose from Cu, Ag, Au, Pt, Sb, Bi, and
Hg. 2. Consider the metals iron and

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silver, both listed in Table 3 on page 286 of the text. Which one

8 Chemical Equations and Reactions

CHAPTER 8 REVIEW Chemical Equations
and Reactions SECTION 3 SHORT

ANSWER Answer the following questions
in the space provided. 1. List four metals
that will not replace hydrogen in an acid.

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Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg.

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