

## Chemistry Chapter 15 Solutions Notes

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### Chemistry Chapter 15 Solutions Notes

Lecture Notes for Chapter 15: Solutions & Solution Chemistry. I. Solutions a. A . solution. is simply a homogeneous mixture i. Homogeneous: same throughout (it does not mean "one") ex: water + sugar, air, alloys, aerated water ii. Heterogeneous means different throughout (does not mean "two") ex: oil + water b. Solute

### Lecture Notes for Chapter 15: Solutions & Solution Chemistry

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Ans: As per the NCERT solutions Chapter 15 Class 12 Chemistry Revision Notes, polymerization is a type of chemical reaction where simpler molecules attach to each other to form a macromolecule. The small molecules are monomers. They bond with each other chemically to form a

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polymer. The best examples are polypropylene, polyethylene, etc.

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Revision Notes on Solution: A solution is a homogeneous mixture of two (or more) substances, the composition of which may vary between certain limits. A solution consisting of two components is called binary solution. The component which is present in large quantity is called solvent and the component which is small in quantity is called solute.

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Solutions Class 12 Notes Chemistry Chapter 2 1. A solution is a homogeneous mixture of two or 9. more chemically non-reacting substances. The components of a solution generally cannot be separated by filtration, settling or centrifuging. 2. A solution may be classified as solid, liquid or a gaseous solution. 3. Solubility is defined as the [...]

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