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Fundamental Of Thermodynamic Van Wylen

In thermodynamics, an isentropic process is an idealized thermodynamic process that is both adiabatic and reversible. The work transfers of the system are frictionless, and there is no net transfer of heat or matter. Such an idealized process is useful in engineering as a model of and basis of comparison for real processes. This is idealized as reversible processes do not occur in reality ...

Isentropic process - Wikipedia

Real gases are often modeled by taking into account their molar weight and molar volume = (+) or alternatively: = Where p is the pressure, T is the temperature, R the ideal gas constant, and V m the molar volume. a and b are parameters that are determined empirically for each gas, but are sometimes estimated from their critical temperature (T c) and critical pressure (p c) using these relations:

Real gas - Wikipedia

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Un gas real, en oposición a un gas ideal, es un gas que exhibe propiedades que no pueden ser explicadas enteramente utilizando la ley de los gases ideales. Para entender el comportamiento de los gases reales, lo siguiente debe ser tomado en cuenta: efectos de compresibilidad; capacidad calorífica específica variable; fuerzas de Van der Waals

Gas real - Wikipedia, la enciclopedia libre

2008 6 2009 1 (OpenCourseWare Consortium, OCWC ...

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On this scale, the unit of thermodynamic temperature T is again the kelvin (K), defined as the fraction 1/273.16 of the thermodynamic temperature of the triple point of water, which is sole ...

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Libro - Wikipedia

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Tentukanlah tekanannya dengan menggunakan (a) persamaan gas ideal, (D) persamaan van der Waals, (c) persamaan Redlich-Kwong, (fl faktor kompresibilitas dan (e) tabel-tabel uap. 2,33 lJap pada 300°C memiliki densitas 7,0 kg/m3.

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